NAVAL SHIP RESEARCH AND DEVELOPMENT CENTER

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ANNAPOLIS DIVISION

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A STATISTICAL ANALYSIS OF HOT-CORROSION TESTS OF SOME EXPERIMENTAL AND COMMERCIAL SUPERALLOYS

By
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H. von E. Doering

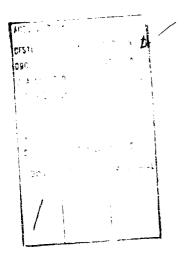
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MATERIALS LABORATORY

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The Naval Ship Research and Development Center is a U.S. Navy center for laboratory effort directed at achieving improved sea and air vehicles. It was formed in March 1967 by merging the David Taylor Model Basin at Carderock, Maryland and the Marine Engineering Laboratory at Annapolis, Maryland.

Naval Ship Research and Development Center Washington, D.C. 20007

A Statistical Analysis of Hot-Corrosion Tests of Some Experimental and Commercial Superalloys

by R. Field, D. J. Fisk, and H. von E. Doering

ABSTRACT

The use of gas turbines in marine power plants depends in part on the development of superalloys which not only possess high temperature mechanical properties but also resist the corrosive effects of sea salt.

As part of a program to develop such alloys, a total of 137 experimental and commercial superalloys, both nickel and cobalt based, were exposed in burner rigs where controlled amounts of sea salt were added to the combustion products of sulfur-containing diesel fuel. Test temperatures ranged from 1600° to 2125° F. Times ranged from 86 to 100 hours with 200 parts per million, and from 489 to 1100 hours with 5 parts per million salt. Corrosion was measured by recording both surface loss and maximum penetration. This experimental work was performed by the General Electric Company under contract to the Naval Ship Research and Development Center.

For each group of alloys tested under similar conditions, a linear regression equation was found that shows the average contribution of each alloying element to the amount of corrosion. The effects of the alloying elements were found to vary with changes in temperature, salt concentration, and whether or not the particular element was part of a simple binary or tertiary alloy, or a complex alloy.

Analyses of variance methods were applied to two sets of factorially designed compositions, one of nickel-base alloys and one of cobalt-base alloys, to determine the possible significance on corrosion of various proportions of single elements and interactions among elements. It was found that in the cobalt alloys significant interactions existed between heat treatment and temperature as well as between heat treatment and chromium content.

ADMINSTRATIVE INFORMATION

This report constitutes Fiscal Year 1969 Milestone 3, on page 98 of the October 1968 Program Summary of the Annapolis Division, Naval Ship Research and Development Center. This work was supported by MATLAB Assignment 1-815-122-A, Sub-project S-F013 06 14, Task 3888, on Gas Turbine Materials, Corrosion.

The results of this study were presented at the Fall Meeting of the Metallurgical Society of the AIME on October 16, 1968.

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NAVAL SHIP RESEARCH AND DEVELOPMENT CENTER

A STATISTICAL ANALYSIS OF HOT-CORROSION TESTS OF SOME EXPERIMENTAL AND COMMERCIAL SUPERALLOYS

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INTRODUCTION

The effects of alloying elements on the not-corrosion resistance of nickel— and cobalt—base superalloys has been a subject of interest to gas turbine manufacturers for a number of years. The application of gas turbines in marine environments has necessitated the development of alloys, for hot section components, which are resistant to the molten salts ingested by the engine. A knowledge of the behavior of alloying elements in either increasing or decreasing corrosion resistance is necessary for future alloy development.

In two recent studies of hot-corrosion resistance of superalloys, 1,2 statistical analysis was employed to establish a multiple linear regression equation relating the weight percent of alloying elements present with the amount of corrosion observed.

It is the purpose of this study to treat statistically the data which was generated for this laboratory in four studies, under contract with the General Electric Company. 3,4,5,6,7,8

Linear regression coefficients and their significance are computed for all alloying elements used in simple experimental (up to four elements) alloys, experimental complex alloys, and commercial alloys. The effects of temperature, the concentration of sea salt, and the type of alloy on the behavior of each element have, where possible, been examined.

Since it was felt that alloying elements do not behave independently but interact, two factorially designed sets of experimental alloys are examined using analysis of variance methods. 9,10

Superscripts refer to similarly numbered entries in Appendix B.

EXPERIMENTAL PROCEDURE

Specimens of all 137 alloys were exposed in a burner rig designed to simulate the environment within the hot section of a gas turbine which was operated while ingesting aerosol sea salt. Figure 1 is a schematic view of the equipment. Diesel fuel containing 1% sulfur was atomized and burned within a ceramic combustion tube. To the flame sea salt was added at either 5 ppm or 200 ppm by weight of air.* Tests with 5 ppm salt were run for 500 to 1000 hours, whereas tests with 200 ppm salt were run for times up to 100 hours only. Thermal cycling was effected by removing the rotating specimen holder and allowing the specimens to cool for 5 minutes every 50 hours during the 500- and 1000-hour tests. The shorter tests were not thermally cycled.

The specimens, nominally 1/8 inch in diameter and 1 5/8 inches in length, were sectioned after exposure; two measurements, surface loss and maximum penetration, were taken as shown in Figure 2.

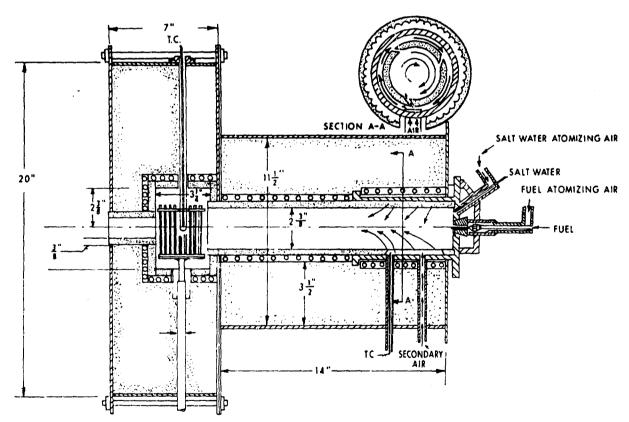
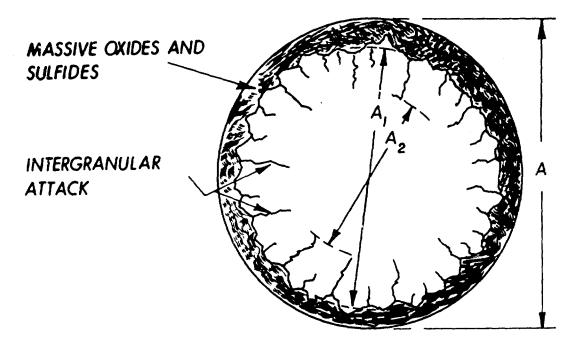


Figure 1
Schematic Cross-Section of Burner Rig

^{*}Abbreviations used in this text are from the GPO Style Manual, 1967, unless otherwise specified.



A = ORIGINAL DIAMETER, MEASURED WITH A MICROMETER.

A₁ = DIAMETER OF STRUCTURALLY USEFUL METAL. MEASURED AT 100X

A₂ = DIAMETER OF METAL UNAFFECTED BY OXIDES AND SULFIDES, MEASURED AT 100X

SURFACE LOSS: A-A1 LOSS IN DIAMETER DUE TO MASSIVE OXIDES AND SULFIDES.

MAXIMUM ATTACK: A-A2 LOSS IN DIAMETER DUE TO ALL FORMS OF OXIDATION AND SULFIDATION.

Figure 2

Method of Measuring Hot-Corrosion Attack

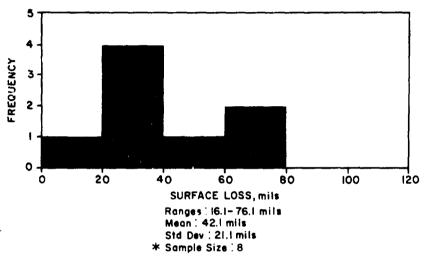
RESULTS AND DISCUSSION

CORROSION MEASUREMENTS

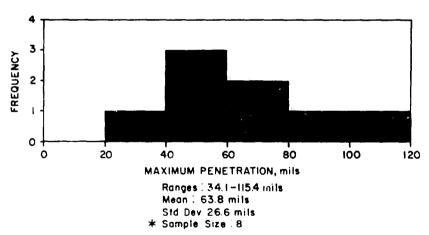
To assess the consistency and accuracy with which corrosion behavior can be measured in the burner rig used in the study, measurements were taken from eight specimens of one heat from an alloy, PA 1 (Heat 1). Each specimen was originally intended as a control to determine the similarity of nominal operating conditions between runs. Therefore, each specimen comes from a different test run, although each test was performed under the same specified conditions of temperature (1750° F), salt concentration (200 ppm) and operating time (100 hours). Thus, the

variance, σ^2 , for the specimens represents dispersion in test conditions as well as variation in behavior of the alloy from specimen to specimen.

A comparison of the data for the surface loss and maximum penetration measurements indicate an average difference of about 20 mils. There is, at the 5% significance level, a significant linear correlation between the surface loss and maximum penetration measurements, see Figure 3.



NOTE: CORRELATION BETWEEN SURFACE LOSS AND MAXIMUM PENETRATION MEASUREMENTS r = 0.8871 (SIGNIFICANT AT 5 % LEVEL)



* TESTED AT 1750° F, FOR 100 HOURS OF OPERATION AND 200 PPM SALT

Figure 3
Frequency Distributions (Surface Loss and Maximum Penetration) for Hot-Corrosion Measurements
Taken of Alloy PA 1 (Heat 1)

Alloy PA 1 (Heat 1) was the only alloy in the study for which the results of a sufficient number of tests performed under similar conditions were available to evaluate the distribution parameters of surface loss and maximum penetration. These estimated measures of dispersion and correlation between surface loss and maximum penetration should not be applied to other alloys, however, or to PA 1 (Heat 1) when they are tested under other conditions.

REGRESSION ANALYSIS

An intensive regression analysis of the 137 alloys included in the development program was completed.

The alloys analyzed included 47 experimental cobalt-base alloys, 73 experimental and 17 commercial nickel-base alloys. test temperatures ranged from 1600° to 2125° F. Times ranged from 36 to 100 hours with 200 ppm salt and from 439 to 1100 hours with 5 ppm salt. A total of 969 tests were examined in the regression analyses. The compositions by weight percent of simple experimental, complex experimental, and commercial nicklebase alloys, plus experimental cobalt-base alloys, are given in Appendix A, Tables 1-A through 4-A. The analyzed composition is given when available, otherwise the nominal is shown. The General Electric Research Laboratory series designated as RL nickel-base alloys are simple experimental alloys, while the Thomson Engineering Laboratory (TEL) series and the Marine Engineering Laboratory nickel (MELNI) alloys are the complex nickel-base alloys. The RL series of the cobalt-base alloys, the Marine Engineering Laboratory cobalt-base (MELCO) series, and the experimental DISCO series comprise the experimental cobalt alloys. The DISCO alloys were intended to be a matrix for dispersion strengthening.

Table 5-A shows the ranges of concentration in weight percent for each element in each group of alloys. The regression equation for the group will be valid only for an element whose concentration lies within the specified interval.

For each group of alloys within a series and tested under similar conditions, a multiple linear regression equation showing the average contribution of each alloying element to the amount of corrosion was found. Tables 6-A and 7-A give the regression coefficients for the equations representing the different groups of alloys at various conditions. Coefficients in Table 6-A are

All tables mentioned in this text will appear in Appendix A.

based on the measurement of surface loss, whereas those in Table 7-A are based on maximum penetration measurements. A positive regression coefficient indicated a tendency for a given element to increase corrosion whereas a negative coefficient indicated a decreased tendency toward corrosion. Other pertinent facts concerning each regression equation are the multiple correlation coefficient, R, and the standard error of estimate, $S_{\rm E}$.

Inspection of Tables 6-A and 7-A indicates the contribution to corrosion of relatively few coefficients with a night level of confidence (95%). The confidence level of many of the coefficients cannot be accurately measured due principally to the few tests conducted of any given alloy. In addition, the coefficients of most elements do not consistently indicate that the element has either a beneficial or detrimental effect. Also, it was not feasible, with the number of tests used in the studies, and to examine all the possible interactions. Tables 3-A through 17-A examine the behavior of each element as it may be affected by test temperature, by time and salt concentration, by whether or not the element is a constituent of a simple or complex alloy, and by the concentration range of the element present.

The effect of temperature in 100-hour tests of the simple nickel-base experimental alloys (RL 1 through 9, and 20 through 51) is shown in Table 3-A. It can be seen that Y, Zr, and Ce are consistently detrimental with respect to maximum penetration at the three temperatures, and that C, Si, Cb, Mo, and W are increasingly detrimental with increasing temperature. On the other hand, Cr, Fe, and Co are consistently beneficial while Al is beneficial with increasing temperatures; Ti is detrimental at 1750° F. The highest degree of confidence (95°) in the above conclusions is indicated for Ti (1675° F), for Zr (1750° F), for Al and Ce (1675° and 1750° F), for Cb, No, W, and Ta (1900° F), and for Cr (all temperatures).

The effects of temperature on the behavior of elements with respect to maximum penetration of the complex experimental nickelbase alloys (MELNI 1, 2, 4, 5, and 7 through 24) in 1000-hour tests are summarized in Table 9-A. It can be seen that Cr, Fe, Co, Zr, Mo, W, and Re have consistent, but not highly significant, detrimental effects and that only Al, Ti, Y, La, and Ta seem to have a beneficial effect on corrosion. Boron seems to promote corrosion with increasing temperature, in contrast to the effects of C and Cb. However, only the effect of Al and Y on corrosion should be considered highly significant (at the 30% confidence level).

In comparison of Tables β -A and β -A, it appears that the effect on maximum penetration of Zr was consistently duplicated

in both the RL alloys for 100 hours and the MELNI alloys for 1000 hours.

Thus, the projection of the behavior of elements in simple alloys under one set of conditions and ranges of concentration cannot predict their behavior in complex alloys at other sets of conditions. In addition, from Table 5-A it can be seen that C, Cr, Co, Zr, and W were present in different concentration ranges in each series of alloys. This fact may have added to the discrepancy.

A more direct comparison between simple and complex alloys is summarized in Table 10-A. Under precisely the same conditions, 1750° F and 200 ppm salt for 100 hours, consistant agreement is found in Al, Ti, Fe, Y, and Zr with respect to maximum penetration. The discrepancy in C, Cr, Co, and W can again be suspected to be due to the differing concentration ranges in both sets of alloys. The beneficial behavior of Mo at 1750° F in the MEINI series in the 100-hour test, in contrast to the 1000-hour test, or in the simple alloys with increasing temperature may indicate that this element contributes to corrosion resistance in a highly sulfidizing environment, but not in a more oxidizing one associated with higher temperatures and lower salt concentrations.

Similar comparisons can be made from Tables 8-A through 10-A with respect to surface loss. In the RL series, the beneficial (or detrimental) influence of C, Ti, Cr, Fe, Co, Cb, and Mo on corrosion was the same for both surface loss and maximum penetration. On the other hand, the influence of B, Y, Zr, and Ce on surface loss was opposite to that for maximum penetration.

A similar lack of agreement exists for the MELNI series at 1600° and 1800° F (Table 9-A). Only C, Fe, Zr, Mo, W, and Re had the same influence on both surface loss and maximum penetration. The disagreement is not surprising in view of the lack of correlation between the two measurements, and thus gives added support for the use of both measurements in hot-corrosion tests.

The test effects of time and salt concentration at 1750° F on the MELNI 1 through 13 series is shown in Table 11-A. It is interesting to note that the effect of time does not change the sign of any regression coefficient although the values do differ. Also, the influence of each element is the same for both surface loss and maximum penetration. A comparison of the effects of the simple RL alloys and the complex TEL and MELNI alloys is given in Table 12-A.

It is interesting to note that none of the effects are consistent, either for surface loss or for maximum penetration. The effects on corrosion for the elements common to the three groups remains the same for both types of measurements.

A similar comparison is made between the TEL alloys and the commercial alloys tested at 1600° F, for 500 hours with 5 ppm salt (Table 13-A). Of the elements common to both series of alloys, all but C, Ti, and Cr show the same beneficial, or detrimental, effect. With the exception of W, all elements had the same effect with respect to both surface loss and maximum attack. With regard to the surface loss of the commercial alloys, the effects of Cr, Mo, and W agree with results of Ryan² for tests performed at 1800° F.

In contrast to the simple nickel-base RL series, the cobalt series (RL 10 through 15 and 52 through 64) in Table 14-A show the alloying elements to have the same type of effect on both surface loss and maximum penetration. Carbon is always detrimental, while Ni, Mo, and W show an increase in corrosion effect with increasing temperatures. Titanium, Cr, Y, Zr, Cb, La, Ce, and Ta show a beneficial effect with increasing temperature.

The behavior of alloying elements in the complex MELCO 1 through 11 series is compared with their behavior in the simple RL series at two temperatures, 1750° and 1900° F, for the 100-hour and 200 ppm salt tests. The elements common to both sets of alloys exhibit a lack of consistent behavior, with the exception of Y, whereby the set of alloys used and the temperature tested each have an effect on surface loss and maximum penetration. Yttrium shows a beneficial effect (sometimes significant, sometimes not) under all the conditions described above.

The effects of temperature on the behavior of elements common to MELCO 3, 4, 5, 7, 8, 9, 10, and 11 can be examined in Table 16-A. Boron is consistently detrimental at all temperatures and in both surface loss and maximum penetration, although the significance is questionable. Chromium, Ni, Ta, and W show reduction in corrosion with increasing temperature, and C, Cu, and Y show inconsistent test results. However, because of the few tests conducted on each alloy, the validity of these conclusions is questionable.

The nature of the tests conducted on MELCO 1 through 11 permits us to examine the behavior of the alloying elements in somewhat more detail than can be done with most of the other alloys examined. In addition to the sign of the regression coefficient, its numerical value may be examined in Tables 17a-A

through 17h-A to determine the sensitivity to corrosion of a given element under two test-times and two temperatures.

Boron

With the exception of 1000-hour tests at 1750° F, B seems to be detrimental with respect to surface loss. Maximum penetration is more dramatically enhanced at 1900° F than surface losses at that temperature, even with the lower salt concentration.

Carbon

The effect of carbon appears to vary most with test conditions at 1900° F, although the significance of the coefficients is questionable.

Chromium

Although Cr is accepted as beneficial in hot corrosion, it appears from the tests that there are cases at the high concentration ranges where it is not. For example, based on maximum penetration values, Cr appears beneficial at both 1750° and 1900° F with the higher salt concentration and 100 hours of operation. However, a longer period of test operation (1000 hours) shows a detrimental effect at a low salt concentration. This seems to indicate that test conditions, oriented more toward oxidizing than sulfidizing conditions, cause Cr to be less helpful.

Nickel

Nickel appears to be the most detrimental at the highest temperature and longest tess even with the low salt concentration.

Copper

Copper appears to be beneficial, expecially at high temperatures and long times, with the exception of the test at 1750° F for 1000 hours.

Yttrium

The effect of Y at all conditions, except the detrimental effect on surface loss in 1000-hour tests, is beneficial. These observations, at least with respect to maximum attack, support the contention that if Y is helpful in oxidation resistance of cobalt-base alloys, then it should be of similar help in corrosion resistance.

Tantalum

In the concentrations present in MELCO 1 through 11, Ta appears to be innocuous (neither beneficial nor detrimental).

Tungsten

Like Ta, W, too appears innocuous or at least inconclusive in its behavior.

ANALYSIS OF VARIANCE

Nickel-Base Alloys

Table 18-A illustrates the testing of three sample specimens of eight nickel-base alloys at each of three temperature levels (a total of nine specimens per alloy).

The analysis of variance (Table 19a-A), using all the given test results for maximum penetration, seems to indicate that only differences in the proportion of chromium used (15% or 20%) made a significant contribution to differences in corrosion measurements.

However, a further examination of the data in Table 18-A indicates that certain test runs (e.g., Nos. 3 and 3) resulted in an exaggerated corrosive effect on certain alloys, as compared to other specimens of the same alloys tested under the same conditions but in different test runs.

Examples in Table 13-A of these particular test specimens include Alloy RL 46, when tested at 1675° and at 1750° F, and Alloys RL 47 and 51, when tested at 1750° F.

Therefore, a new analysis of variance was performed (Table 19b-A) in which Tests 3, 3, and 13 were eliminated; the remaining test measurements for each alloy obtained under similar conditions were averaged together. In the new analysis, besides the differences in corrosion due to differences in chromium, those concerning titanium and temperature levels also seemed to make significant contributions to corrosive behavior.

It should be noted that variations in the proportion of nickel used in these nickel-base alloys, although not specifically listed in Table 13-A, may also contribute to an increase or decrease in corrosive effects.

Cobalt-Base Alloys

Hot-corrosion tests were performed on two specimens each of nine cobalt-base alloys, at each of three temperature levels, as shown in Table 20-A. For each two specimens, one specimen was placed in the test chamber as cast, while the other specimen was given a special heat treatment before being placed in the test chamber.

An analysis of variance using the maximum penetration measurement is presented in Table 21-A to determine the relative effect on corrosion of heat treatments or lack of heat treatments versus the proportions of Chromium, nickel, and tantalum used within each alloy. The results of these analyses are presented graphically as well as in analysis-of-variance tables, according to statistical methods described by Hoel⁹ and Brownlee.¹⁰

Taking into consideration all the given factors, there seems to be considerable interaction between (H) (heat treatment or lack of heat treatment) and the test temperatures used (T), as well as between (H) and the proportion of chromium used in an alloy (Cr). A significant interaction indicates that a specific combination of factors may possibly affect the outcome of a test in a somewhat different fashion than would each of the factors considered by itself (called a main effect). This analysis also indicates a possible interaction between test temperatures and the proportion of chromium used, as well as significant main effects of temperature and nickel considered independently.

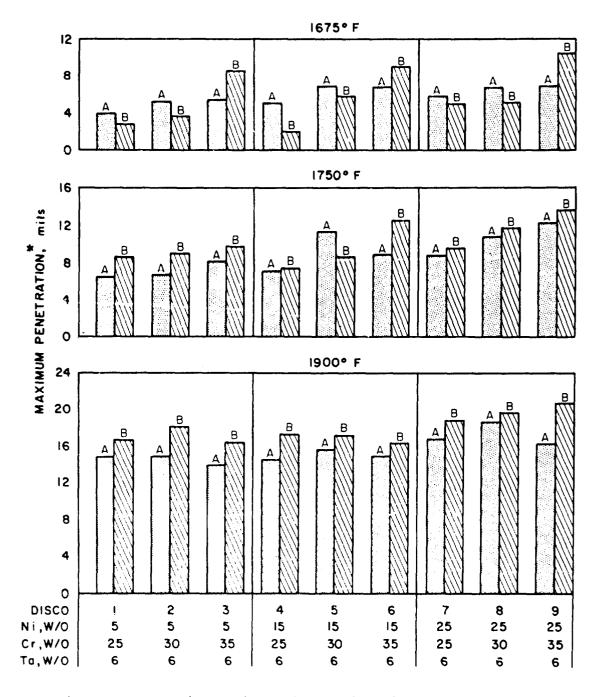
Because of the highly significant effect of temperature on most of the other factors, a separate analysis of variance was performed in reference to each separate temperature level (1675°, 1750°, and 1900° F). The result of this analysis is also presented in Table 21-A and shown graphically in Figures 4 and 5.

As can be readily seen on examining Figures 4 and 5, there is an increase in corrosion due to higher test temperatures for all proportions of nickel and chromium used in the alloys and in respect to both the heat-treated (B) and nonheat-treated specimens (A).

At an operating temperature of 1900° F, heat-treated specimens (B) show a consistent increase in corrosion over nonheat-treated specimens (A). At 1750° F, one set of specimens out of nine sets shows a reverse relationship (a decrease in corrosion for the heat-treated specimen), while heat-treated specimens tested at 1675° F show about a 50-50 chance of either increasing or decreasing the corrosive effect.

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A = Nonheat-treated B = Heat treated

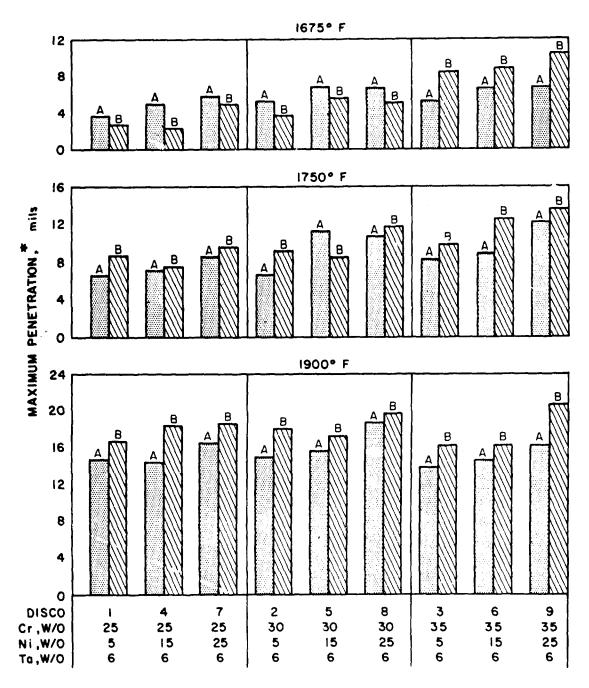


^{*}Maximum penetration values from Table 20-A.

Figure 4
Hot Corrosion of DISCO Alloys Arranged by Nickel Content

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A = Nonheat-treated
B = Heat treated



^{*}Maximum penetration values from Table 20-A.

Figure 5
Hot Corrosion of DISCO Alloys Arranged by Chromium Content

In a similar fashion, the general increase in corrosion test results due to increases in nickel content (Figure 4) at each temperature level is evident, as is also the general decrease in corrosion measurements due to increases in chromium content (Figure 5) at each temperature level. However, an increase (or decrease) in nickel content has less apparent effect on corrosion when tested at 1900° F, while a change in chromium content has less apparent effect on corrosion when tested at 1675° F. The effects of nickel, chromium, and temperature on corrosion test results may thus be considered main effects at various temperature levels. As shown in Table 21-A, nickel seems to be very significant at all three temperatures, while chromium is most significant at 1750° F and least significant at 1675° F.

Thus, the analysis of variance presented in Table 21-A and the graphical representations in Figures 4 and 5 indicate that heat treatment has no significant effect on corrosion when specimens are tested for 100 hours at 1675° F, a somewhat greater significance when tested at 1750° F, and a very significant effect when tested at 1900° F. However, the direction of change (increase or decrease in test results) versus the change in test conditions is not presented in an analysis of variance, but must be determined by further examination of the data.

In addition to changes in corrosion test results due to changes in chromium and nickel content, the experimenter should not neglect the possible effects of tantalum (6%) and the increases or decreases in cobalt content which depend on the total proportion of other metals used in each alloy.

SUMMARY AND CONCLUSIONS

The hot-corrosion behavior of a number of experimental and commercial nickel- and cobalt-base superalloys was statistically analyzed. Tests were made in a burner rig using diesel fuel to which controlled quantities of sea salt were added. The significant results of the study are as follows:

- In general no one-to-one correspondence could be found between surface loss and maximum penetration for all the alloys tested under various conditions. Therefore, in studies having such a spread of alloy compositions, both measurements are recommended.
- Projection of the behavior of elements in simple alloys under one set of conditions and ranges of concentration cannot always predict their behavior in complex alloys at other sets of conditions.

- Although Cr is considered a beneficial constituent, this analysis indicates that additions greater than about 20% in nickel-base MEINI alloys had a tendency to increase corrosion slightly.
- The lack of consistency in the effect of various elements on corrosion indicates the possible existence of interactions among elements as well as the need for testing a greater number of specimens of any given alloy.
- In a series of cobalt-base DISCO alloys with factorially designed compositions, it was found that there were significant interactions between Cr and temperature, between heat treatment and Cr, and between heat treatment and temperatures.
- Alloying elements, which show effects of doubtful significance, should be investigated by designed experiments, using a sufficient number of replications.

Appendix A

Tables

Table 1-A
Chemical Composition of Simple Experimental Nickel-Base A_Loys*

RL										E1-	emei	nt,	X,									T E
Desig-		С	Al	Si	Ti	٧Ī	Cr	Mn	Fe		Ni		Y	Zr	Cb	Мо	La	Ce	H£	Ta	W	R
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40	<u> </u>	0.02			└	+-			10.0	. 	+	+-			1.00	-	+	1			1	7
41		0.02				┿	15.0		20.0		+	+		 	 -	 	+-	 	1		1	T
42		0.02			 -	┿	15.0		1,0.0	Ή	+-	+		0.500	 	 	+-	 		1		\top
113	0.150			<u> </u>	4.50	+	15.0		┼──	+	+-		 	+	 		1					\Box
44		0.02	2.50	ļ	3.00		15.0		+	+	+-	+-	 	 		1	1	1				\Box
45	 		2.50		4.50		15.			+	+-	+-	 	 	 				T.			\perp
46	-		5.00		7.00	_	15.0		+	+	+	+	-									\perp
47	 		5.00		74.50		50.0		+	+	+-	+-	 -	1		1	T^-				<u> </u>	
49			2.50		7.00		20.0		+	+	+-	+-	\vdash			1	1			l	L	\perp
49	 -		2.50		14.47		20.0			 		+-	+	 		1	1	I				\perp
50 51	↓		5.00		7.00		20.		+	+	+-	+-	1	1	1			1			1	_[

*Composition in weight percent. B = Balance

Table 2-A

Chemical Composition of Complex Experimental Nickel-Base Alloys*

Desig-										Ī	ller	nent	, %									
nation	В	С	Al	Si	Тi	V	Cr	Mn	Fe			Cu	Y	Zr	Cb	Мо	La	Ce	Hf	Ta	W	Re
			·			- 7 -		,			TEL										11	
1		0.12			2.00		5.0			10.0		4-4			0.50			 	1-1		4.00	
2	ļ	0.14			2.00		0.0			10.0		\perp			0.50			↓_	1		4.00	
3		0.14			1.65		5.2			10.0		1		 	0.51			┦	├		4.14	
4		0.15			2.00		0.1			10.0		1		ļ	0.50			٠.	 		4.00	
5		0.16			0.02		5.1			10.0				}	0.50			 	11		4.00	ļ
6		0.17			2.96		5.0			10.0					0.50		<u> </u>	<u> </u>	1		4.00	
7		0.12			3.96		5.0			10.0		\Box			0.50						4.00	L
3		0.11			2.00		5.0			10.0					0.50			<u> </u>			7.49	
9		0.12			2.00		5.0			10.0	! _			<u> </u>	0.50			L			1.77	
10		0.15			2.00		5.0								0.50						4.00	
11		0.15			1.93		5.1			24.2	Ĺ	Ш			0.50						4.73	
12		0.13	6.20	\perp	3.93		5.0			0.1					0.50	1.7		Ι				
											ELN	I										
	0.009				1.30		8.7			14.9	<u> </u>			0.092		4.0						
2	0.009	0.0)	3.50		1.80		9.3		0.2	14.3				0.081				I			I	4.5
	0.006				2.03		0.3			15.3				0.059		0.1					T	
	0.009				1.32		3.7			14.9				0.095		4.0						
	E00.0				2,16		₹.5			23,4			0.200	0.049		3.9						
	0.009				1.32		3.3		0.2	15.2				0.091		4.1	80.0					
	0.015				3.12		4.8			11.3				0.038			0.25				6.00	
	0.015				2.66		1.5			13.8				0.084			0.26				6.10	
	0.009				4.13		2.0			13.0				0.063			0.12			2,40	5.90	
	0.015				3.54		4.5			12.3	Γ.	\Box		0.083			0.31				5.90	
	0.014				3.10		5.0			12.3				0.084			0.17				0.00	
12	0.015	0.05	2.02		3.10		4.3			12.3				0.093			0.67				6.00	
13	0.017				3.10	5	4.4			12.7				0.031			1.45				5.90	
12;	0.021	0.14	2.26		1 . 55	1	9.3			9.4				0.100			0.16				5.65	
17	0.021	0.14	2.23		3.93	1	5.3			9.4				0.015		2.0	0.17	T		1.30	1.95	
1)	0.024	0.14	4.03		1.75	1	9.7			9.4				0.140		i	0.17	T			5.75	
	0.017				4.40		7.5			3.6				0.004		3.9	0.17	1				
	0.021				2.34	1 1	3.3			7.9				0.160		2.2	0.16				2.05	
	0.012				2,00	1	5.2							0.120	2.10	3.3	0.12				3.20	
	0.023				1.37		रें उ			7.6	_	 - 		0.000						1 35	1.30	

*Composition in wordht percent. BA = Balance

Table 3-A Chemical Composition of Commercial Nickel-Base Alloys*

		,			· · · · · ·					ent,			_									<u> </u>
Designation	В	С	A1	Si	Ti	V	Cr	Mn	Fe	Со	Ni	Cu	Y	Zr	СЪ	Мо	La	Ce	Нf	Ta	W	Ţ
ams 5397	0.014	0.18	5.50		4.70	1.00	10.0			15.0	ВА			0.060		3.0						
PA 2	0.015	0.14	4.86		1.80		8.8		0.3	10.2					1.20						12.30	ŗ
AMS 5391A (Heat 1)	0.007	0.12	5.69		1.01		13.4		1.7						2.37	3.7				0.17		
ams 5399		0.08	1.60		3.15		19.1			11.1	·					9.9						İ
AMS 5384 (Heat 1)	0.005	0.08	3.02		3.05		19.4									3.9						
AMS 5384 + Ce	0.005	0.08	2.97		2.98		18.8		0.2	19.4						4.2		0.50	-			T
AMS 5384 (Heat 2)	0.006	0.07	3.00		3.00		18.5			18.5						4.0						
PA 7 (Heat 1)	0.013	0.12	4.70		3.63		14.8			18.9			-	0.050		5.0						+
PA 7 (Heat 2)	0.027	0.09	4. 3 5		3.18		15.2			18.6						5.0						!
PA 7 (Heat 3)	0.015	0.12	4.28		3.48		14.6			15.2						4.1						
PA 7 (Heat 4)	0.015	0.12	4.27		3.41		14.3			15.1						4.2						
PA 7 + 0.5 Ce	0.020	0.07	4.30		3.3 5		14.3			15.0				0.030		4.2		0.50				
PA 7 + 0.2 Ce	0.020	0.07	4.30		3.35		14.8			15.0			!	ı		4.2		0.20				_
PA 1 (Heat 1)	0.014	0.10	4.43		2.07		15.2		0.2	24.3				:	.	4.6		1			-	_
PA 1 (Heat 2)	0.014	0.03	4.50		2.51		15.0			25.3			-			4.5			;	,		
PA 5	0.015	0.07	5.40		2.50		11.0			15.0	:		+		0.45	6.5			i		1.50	_
PA 6	0.013	0.05	6.30		0.10		16.7						1	0.100	1.00		1			1.90	1.95	-

*Composition in weight percent.
PA = Proprietory alloy
BA = Balance

Table 4-A

Chemical Composition of Experimental Cobalt-Base Alloys*

esig-	-					,						ei.en		,							
tion	В	С	Al	Si	Ti	V.	Cr	Mn	Fe	Co	Ni	Cu	Y	Zr	Cb	Мо	La	Ce	H£	Ta	W
											,	ST.									
10		0.01			T	М	25.6	Γ	T	BA	Τ	<u> </u>						T			
11		0.02	_			П				1			1								
12		0.03				Н	25.7		 		<u> </u>		 -								7.88
13		0.01			 		26.2	<u> </u>	_	_						6.0					
14		0.02				П	25.0			1										7.03	
15		0.02				П	26.4			1	10.2		1								
2		0.45				П	15.0			1	10.0										
53 54		0.45					25.0				10.0										
54		0.45					35.0				10.0										
55 56		0.45					25.0				10.0		I							8.00	
6ز		0.45					25.0	l			10.0									15.00	
57		0.45				П	25.0	1		Γ	10.0										8.00
58		0.45					25.0				10.0										15.00
9		0.45			4.00		25.0				10.0										
50		0.45					25.0				10.0		0.150								
51		0.45					25.0				10.0						0.15				
2		0.45					25.0				10.0							1.00			
53		0.45				П	25.0				10.0			0.500							
54		0.45					25.0				10.0				4.00						
											М	LCO									
1	0.015	0.23			T	Π	27.8		Τ	\Box	10.7		1								7.00
5	0.012	0.25					30.0				0.5		0.150								8.10
3	0:012	0.26					30.7				9.7									1.80	7.80
4	0.014	0.27					29.3				9.8									3.45	7.80
5	0.013	0.26				П	29.8				9.8		0.130							3.60	8.00
6	0.012	0.40				П	29.8				9.8		0.190							3.65	8.10
7	0.013						33.0				9.3		0.180							3.80	
3	0.012						34.7				9.5		0.170							4.00	
9	600.0						31.8				9.9	2.95	0.130								7.85
10		0.42					31.8				9.9		0.010								7.85
11	0.013						30.5				9.9		0.120							2.85	8.00
12	0.018			0.23			23.9				10.4		0.250							3.10	
13	0.013			0.24			23.3				10.0		0.200	0.150					L	3.45	7.80
14	0.016			0.21			23.7				10.7		0.180			L			0.15	3.10	
15	0.017			0.23			23.6				11.1		0.550			\sqcup			$\sqcup \sqcup$	3.40	
16	0.015			0.27			28.5			<u> </u>	20.2		0.110							3.50	
17	0.014			0.13		Ц	28.5	0.31	0.2		10.3		0.130								10.50
13	0.017	0.45	لــــا	0.22	L	Ш	23.8	0.09	10.2		10.7		0.110		L				L	3.10	7.40
											ום	sco									
1							25.9			BA	5.2									5.77	
5	.	0.01	<u> </u>		ļ	\sqcup	31.0		 	-	5.2	L				 			<u> </u>	6.00	
3		0.01	L_		<u> </u>		35.?		1		5.2								 	6.00	
I ₄			-		L		25.3		L	_	14.3		ļ			└			L	6.00	
4,		0.01	<u></u>		<u></u>		30.9		ļ	ļ	14.6					 _				6.00	
ϵ_{-}		0.01					3€.3		ļ		14.7			ļ		 				6.00	
7		0.02		0.30	1		25.9	0.10	<u> </u>		23.3				L	$oxed{oxed}$				5.35	
3		0.04					31.0				24.2									6.00	
)		0.03					35.9				24.6									6.00	l
1.3		0.01	1	_		i T	25.6		1	i -	14.3		0.290			· 1	T T		1	6.00	

*Composition in weight p BA = Balance

Alloy	В	C	Al	Si	Ti	V	Cr	Mn	I
_									
RL 1-9, 20-51		0.01-0.04	0.0-9.99	0.0-2.00	0.0-9.00		0.0-25.5		(
TEL 1-10		0.11-0.17	2.65-6.16		0.02-3.96		10.1-25.0		
TEL 1-12		0.11-0.17	2.65-6.20	j	0.02-3.93	j	10.1-25.0		
MELNI 1-13	0.006-0 017	0.05-0.10	1.34-3.68		1.30-4.13		13.3-25.0		
MELNI 1-24	0.006-0.024	0.05-0.14	1.34-4.12		1.30-4.40		13.3.25.0	1	
MELNI 1, 2, 4, 5, 7-24	0.003-0.024	0.05-0.14	1.34-4.12		1.30-4.40		13.3-25.0		١,
Commercials	0.0-0.027	0.05-0.13	1.60-6.30		0.10-4.70	0.0-1.00	3.3-19.4		
								<u> </u>	
						 			
RL 10-15, 52-64	0.01-0.45				0.0-4.00		0.0-35.0		
MELCO 1-11	0.003-0.015	0.83-0.40					27.3-34.7		
MELCO 3-5, 7-11	0.0-0.013	0.26-0.42				-	29.3-34.7		
MELCO 2-11	0.0-0.014	0.26-0.42		0.0-0.27			29.8-34.7		
MELCO 3-5, 7-18	0.0-0.013	0.26-0.61					23.5-34.7	0.0-0.62	
MELCO 1-13	0.0-0.013	0.23-0.61		0.0-0.27			27.3-34.7	0.0-0.62	,
		j	ļ						

		Timo	Sea Salt									
Alloy	T °F	hour		В	C	Al	Si	Ti	T V	Cr	Mn	Fe
	 -	1	I PP	<u> </u>	·	J	1	<u>:=</u>				1
RL 1-9, 20-51	1675	100	500	+130.0 €	1-265.0	cl +4.73 a	-6.33c	-1.73 b	Τ	1 -3.66 al		-0.
	1750					c +5.50 a		+0.3110		-2.52 a		-0.
	1900	100	200	+152.6 c	+696.0	c +1.24 c	-1.17c	-0.735c	 	-3.13 a		-0.
TEL 1-10	1600	100				1		1	† 	1		
	1750				+614.0	c -29.6 c		-23.0 c	 	-2.54		
	1600		5			c -9.61 c		-7.95 c		-3.43 ы		
TEL 1-12	1750		200			c -4.25 c		-9.25 c		-4.63 a		
	1600		5			a +20.9 a		+10.2 c		-4.15 a		
		1100	5			c +1.79 c		-2.50 c	·	-3.9€ a		
MELNI 1-13	1750			+3249.0 0	-112.0	c +12.9 c		+13.2 c		+5.42 c		-54.0
		1000	5	+6063.5 °		+32.0 c		+34.2 c		+14.6		-2062.
MELNI 1-24		1000	5	-233.0 4		c -1.79 E		-0.302 _C		-0.437 c		-3.0
MELNI 1,2,4,5,7-24			5	+33.7 °		-0.519		+0.753c	<u> </u>	+0.153c		+21.3
		1000	5	_+59.9 [▽]		+0.737		+0.77 ¹ 4c		-0.055c		+3.
Commercials	[1600	500	5	-1103.0 c	+99.5	c[+70.0 c		-6,66 c	+33.9c	+4.73		-9.9
RL 10-15, 52-C1	1675		200		+47.9	b	T	-6.49 a		[-4.65 a]		Γ
	1750		200		47.4	c		-6.44 a	1	-4.64 a		
	1)00		800		+33.0	e e		-4.32 c		-2.30 5		
MELCO I-II	1750	100	200	+465.0 0	+1.30	c			·	+0.213c		
	1900	100	500	+556.0 (+3.46	c	† 	†	 	+0.417c		
		1000	5	-163.0 °	-3.01	7	1		 	+0.021		ļ ———
		1000	5	+445.0 °	-15.1	C			 	-0.742b		
	1600		5	-17.6 c	-1.30	ь		 	1	-0.021		
MELCO 3-5, 7-11	1600		5	+19.3 c		C			 	+0.0220		
	2000		5	c 0. رورو+	+15.0	С	1	 	†	-0.3200		
	2125	500	5	+13/1.0		c				-10.7		· · · · · · · · ·
METCO 5-11	2050		5	+353.0 0		a	1			+0.3465		
MELCO 3-5, 7-13	2125		5	+11511.0		C	-150).O a	·		-0.038 c H	-117.0 c	+416.
MELCO 1-13	1 100	1000	5	#413.00 C	+0.933	c	-150 F 10			0.133	+1 1,50	+20.
a -) confidence	leve	1. b .	- 70 7	confidence	level c	- Confide	ngo lough	i		<u> </u>		+ 1000

a - M confidence level, b - M confidence level, c - Confidence level is equal to or greater than 60 but less T - Temperature, R - Multiple correlation coefficient, S_E - Standard error of estimate, N - Number of tests per re

A

					Eleme							Ce	
	Cr	Mn	Fe	Co	Ni	Cu	У	2r	Ср	Mo	La		<u> </u>
				Nickel Base	Alloys								,
	0.0-25.5	;	0.0-20.0	0.0-10.0			0.0-0.15	`0.0-0.50	0.0-7.00	0.0-6.01		0.0-1.0	
Ì	10.1-25.0			0.0-10.0		1			0.50-0.51	0.1-2.2		1	
[10.1-25.0		1	0.0-24.2					0.50-0.51	0.1-2.2			
	13.3-25.0		0.0-0.2	11.3-23.4			0.0-0.200	0.063-0.095		0.0-4.1	0.0-1.45		
	13.3.25.0		0.0-0.2	0.0-23.4			0.0-0.200	0.015-0.160	0.0-2.10	0.0-4.1	0.0-1.45		
1	19.3-25.0	Ì	0.0-0.2	0.0-23.4		·	0.0-0.200	0.015-0.160	0.0-2.10	0.0-4.1	0.0-1.45	1	
00.1	3.3-19.4		0.0-1.7	0.0-25.3				0.0-0.100	0.0-2.37	0.0-6.5		0.0-0.50	1
	<u>. </u>											<u> </u>	L
				Cobalt Base	Alloys					· • · · · · · · · · · · · · · · · · · ·			
	0.0-35.0				0.0-10.2		0.0-0.150	0.0-0.500	0.0-4.00	0.0-6.0	0.0-0.15	0.0-1.00	
	27.3-34.7		:		0.5-10.7	0.0-2.95	0.0-0.190						
	29.3-34.7	ļ			9.3-9.9	0.0-2.95	0.0-0.130						
	29.B- <i>3</i> 1.7				0,5-0,1	0.0-2.95	0.0-0.190		}				
	24.14-54.7	0.0-0.61	0.0-0.3		1.3-20.2	0.0-2.95	0.0-0.550	0.0-0.105					
	77.3-51.7	0,0-0.62	0.0-0.3		0.5-20.2	0.0-2.95	0.0-0.550	0.0-0.105					
	j	ļ						l				}	1

E.														
L					Elemer	nt								
V	Cr	Mn	Fe	Co	Ni	Cu	Y	Zr	Ср	Mo	La	Ce	IIE	Ta
					cel Base A	lloys					,			
L] -3.66 a		-0.707c	-0.4920			-77.3 c	-79.6 c	1,69 c	-2.54 c		-11.7 c		-1.17
	-7.5% a		-∩.053c	-0.005c			-2.69 c	-1.13c	+1.56 °	+0.327 c		-1.07°	<u> </u>	+0.059
	-3.13 a		-0.007c	-0.617c			-34.7 c	-71.4 c	+1/4.5 a	+2.47 c		-12.7 C		+1.19
	<u> </u>		-								ļ	 		
	-2		.l	+€.73 c					-1363.0 c	-56.7 c		J		<u> </u>
	-3.43 D		1 7	+0.3010	T		T		-1922.0 c	+21.6 c	l			
	- 1.03 a			+0.1 <i>5</i> 0 c					+57.0 c	+).44 c				
	-4.1; a			-0.4200					-931.0	+10.7 c	[
	-3. E a			-0.023c					+2134.0 c	+17.5c c				
	- 44. C		4. i C	+1.71 a			+1.2.1 b	±533.0 €		-3.27 b	-9.33 c			+11.7
	1-1/1		-8062.7 c	+5.4 0	1		+35.3 c	+15€0.5 c		-6.37 c	-24.1 c			+29.8
	-0.437		-3.10 cl	+0.1116			-6.01 c	±20.2 c	+1.27 a	-0.420 c	+0.2100			-0.712
	10.13		+21.2	-0.03%			-2.05 c	+2. 55 €	+0.3336	+1.54 c	+0.352c			+0.336
	-3.055		ो - मोस्टल	-0.0930	 		45,05 C	+3.54 c	+0.632c	+1.1/1 c	-0.7240			+0.613
3.10	H. 73 .		-1.17					-1140.0 a	-103.0 c	+2.23 c		+3.460		1-56.5
														
				Caba	It Base Al	1:175				,		, , , , , , , , , , , , , , , , , , , 		
	<u> </u>				-0.(00c		130.0 a	-5/4 /4 a	-6.72 a			-27.1 a		-2.10
	1	T		T	0 1 30		173 0	17.0	(65 0	L 4/L 07 >	-2676 0 3	1-27 2 5		-2.03

	Cobalt Base Alloys	
-9-27	-0.00c -130.0 $-9h.h$ -6.72 -1.01 -2713.9 -27.1 -2.1	
-4.14	-0.16 $-173.0 0.00$ $-174.0 0.00$ $-6.65 0.00$ $-14.07 0.00$ $-2676.0 0.00$ $-27.2 0.00$	13_1
	+0.031c -127.0 c -37.5 c -4.04 c $+21.5$ a -1335.0 c -10.1 c -1.3	57
1.0.1131	40.5 Me -2.12 -0.323 -1.6	
10.50	-0.607c 40.053c -33.0 a	· 37 c
10.00	+0.0°0e -0.177e +0.01 e	777
	+1.52 引 -3.25 引 +8.27 引 -2.6	
	-0.700c +0.027c +0.350b +0.0	400
40.000 (40.00 40.00% 40.	
-2, % \(\text{S}\)	-0.2 40.0/16 -31.) +0.2	1120
1987 1	-15.1 1-15.3 dHr.1 d	
+0,3400	-0.179c -0.133c -0.04 a +0.9	
-0.0% +117.5 + +015.5 ·	-197.0c -10.0) (
-0.5 (1) 4 4 5.5 (1) 45.5.5	48.7715 -2.16 A +0.435 A +25.1 A	11 2
Communication than the first feet than a		
te, N = Namber of the two seasons were a	analysis, 4 - here rept value for redression equation	1
■ () _~		
*		



		,	Ce			W	Re
Cb	Mo	La	L	H£	Та	w	, Re
							,
-7. 00	0.0-6.01		0.0-1.0		0.0-12.0	0.0-24.0	
50- 0.51	0.1-2.2					1.77-7.49	
0-0.51	0.1-2.2					1.77-7.49	
	0.0-4.1	0.0-1.45			0.0-2.40	0.0-6.10	0.0-4.52
-2.10	0.0-4.1	0.0-1.45			0.0-2.40	0.0-6.10	0.0-4.52
-2.10	0.0-4.1	0.0-1.45			0.0-2.40	0.0-6.10	0.0-4.52
)- 2.37	0.0-6.5	ı	0.0-0.50		0.0-1.90	0.0-12.3	
-4.00	0.0-6.0	0.0-0.15	0.0-1.00	-	0.0-15.00	0.0-15.00	
					0.0-4.00	6.30-3.10	
					0.0-4.00	6.30-3.25	
					0.0-4.00	6.30-3.25	
				0.0-0.15	0.0-4.00	6.30-10.50	
				0.0-0.15	0.0-4.00	6.30-10.50	
					!		

0	La	Ce	Ηf	Ta	W	Re	R	s_{E}	N	1
			· · ·							
5/4 c	7	-11.7 €		-1.17 c				26.9	111	72.5
827		-1.07°		+0.059c			0.53	25.6	111	35.2
47		-12.7 €		+1.19 c	+3.27 a		0.67	26.1	103	32.7
7 0	·				-25.7c ⊂		0.36	22.3	12	1216.0
6				Ĺ	+5.22 c		0.35	12.9	20	994.0
44 (-3.02 c		0.33	16.2	16	161.↑
7(-3.33 c		0.35	20.2	23	533.0
5 c (+0.366c		0.91	14.1		-1003.0
27				+11.7 c		ļ	0.93	0.67	31	-251.0 -7.4
87 (-27.3 c	-39.0 c		0.33	0.44	13 40	13.6
420	+0.2100			-0.712c				0.61	12	-10.7
5/; 1/1	+0.3%			+0.336c			0.96	0.50	25	-10.7
	-0.75%			+0.613 _e		+1.70 C	0.39	26.3	45	-91.7
2 당 :	<u>'</u>	+3.410		1-56.5	-0.31(c	L	0.59	10.5	116	-112./
O 1	T-2713. L	-17.1		-2.10 a			0.33	14.5		123.0
07 :	<u>∃-267(.0</u> a	-27.1		-2.03 a			0.36	16.2	74-3	123.0
5.	-1333.0	-11.1 0		-1. <i>5</i> 7_c			0.38	24.2	40	74.1
		1 1			+1.41 c		0.92	1.69	11	-13.9
				+0.437 c			0.35	1.43	15	1.30
				+0.077c			0.96	0.290	11	2.76
		· ·		-2.31 n	-2.33 a		0.32	2.60	23	47.9
	ļ				+0.0450		0.33	0.031	11	1.03
		<u> </u>			+0.011		1.0	0.013	3	_1; 7;1;
		1			-0.2550		0.10	1.73	16	+75.3
	 			-13.) c		ļ	0.096	2.00	3	+1073.0
	 		1 1/2 5		+1.52 a		0.05	1.04	50	-43.2
-	+	ļ		-10.0 c			0.01	5.22	42	- 353.0 +15.7
	4	لحد حد د	-19.60	<u>: -1.01.0</u>	-0.1240	<u> </u>	10.00	1_(1)	1 40	±10.1

Table 5-A

Ranges for Each Alloying Element

Table 6-A

Regression Coefficients for Surface Loss

MATLAB 243

A-

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	T											_
		mim-	Sea									
711	T °F	Time	1				C :		<u>v</u>		r - a -	-
Alloy	F	hour	ppm	В	С	Al	Si	Ti		Cr	Mn	1_
RL 1-9, 20-51	1675		200		c -117.0 c					-3.61 a		
	1750	100	200	-883.4	a +322.0 c	+3.87 a	+1.69c			-3 .90 a		
	1900		200	-457.0	c +728.0 c	-0.772c	+19.1 b	-1.47 b		-3.10 a		Γ
TEL 1-10	1600		200		+345.0	-13.9 c		-7.15 c		-2.24 c		Γ
	1750		200		+330.0	-27.1 c		-27.1 c		-8.12 c		Г
	16° 3	500	5			+3.10 c		-2.02 c		-5.70 a		Γ
TEL 1-12	1750	100	200		-448.0 c	-3.56 c		-13.1 c		-9.21 a		
	1600	500	5			+23.5 a		-10.2 b		-6.07 a		Γ
	1750	1100	5		-87.0	+1.75 c		+0.437 c		-4.70 a		
MELNI 1-13	1750	100	500	+1583.0	c -59.6 c	+12.0 c		+17.2 c		+5.47 c		Γ
	1750	1000	5		c -302.9 c			+72.2 c		+31.81 c		۲.
MELNI 1-24	1600	1 000	5		c +50.1 c			-3.62 c		+1.14 c		
MELNI 1, 2, 4, 5, 7-24	1600	1000	5	-512.0	c +108.0 c	-3.85 c		-1,25 c		+3.14 c		-
, , , , , ,	1800	1000	5		c -5.20 c			-4.48 c		+1.44 c		
Commercials	1600		5		c +408.0 c			-8.27 c	+14.5c	+7.14 c		1
					· •	ا ــــــــــــــــــــــــــــــــــــ				· · · · · · · · · · · · · · · · · · ·		٠
RL 10-15, 52-64	1675	100	500	· · · · · · · · · · · · · · · · · · ·	+51.3 a	1		-6.43 a		-4.43 a		1
	1750		200		+47.0 E	,		-6.37 a		-4.40 a		1
i	1900		200		+37.3			-3.55 c		-2.53 a		†-
MELÇO 1-11	1750	100	200	- 796.0						-2.67 c		1
	1900	100	200	+1094.0	c +29.2 c	:				-1.14 c		T
	1750	1000	5	+76.0						+0.533c		r
ļ	1900	1000	5	+1320.0						+0.583c		Γ
1	1600	500	5	+484.0	c +2.71c					+0.156c		1
	1600	500	5	+434.0	c -6.60 c					+1.55 c		_
MELCO 3-5, 7-11												_
MELCO 3-5, 7-11	2050	500	5	+373.0			į į	į į	- 1	-2.23 c	!	
MELCO 3-5, 7-11	2050 2125			+378.0 +579.0	c -52.6 c					-2.23 c		-
MELCO 2-11		500	5		c -52.6 c +6.34 c							
	2125	500 500	5 5	+579.0	c -52.6 c c +6.34 c a +27.0 c		-737.0 c			-10.7 c		



Table 7-A Regression Coefficients for Maximum Penetration

3.57 al	Si	Ti								ent			
3.57 1			V	Cr	Mn	Fe	\mathbf{I}_{-}	Co	Ni	Cu	Y	Zr	Cb
3.57 1													
•າ. ໆ / ຈ !	-1.24c		,	1 2 () -1		0.075			Base Alloy	/S	+76.5 c	+61.6	-1.9
3.87 a	+1.69c	-2.01 a +0.309c		-3.61 a		-0.867 -0.445					+170.0 c		
0.772c	+19.1 b	-1.47 b		-3.10 a		-0.445 -0.055					133 5 0	+127.0	
3.9 ci	+19.1 b	-7.15 c		-2.24 c		-0.055		003 c			+ 77.7 C	-1E(1.0)	-320.
7.1 c		-27.1 c		-3.12 c				95c c					-2352.
3.10 c		-2.02 c		-5.70 a				565 c			 		-2522.
3.56 cl		-2.02 C		-9.21 a		·		445 c					-1511.
3.5 a		-10.2 b		-6.07 a				303 0			 		-1925.
1.75 c		+0.437 c	<u> </u>	-4.70				0006c			 		+3054.
2.0 c		+17.2 c		+5.47 c		-39.7	c +2.				+67 3 c	+705.0	
3.5 c		+72.2 c		+31.31 c			c +3.					+2731.3	
7.00 c		-3.62 c		+1.14 c			c +1.					+151.0	
3.35 c		-1,25 c		+3.14 c			c +0.				-54.3 c		
6.64 1		-4.43 c		+1.44 c				249 c			- 37.1 b		
1.0 c		-3.27 c						33b b				-)74.0	
							<u></u>				······································		
							Co	balt-	Base Allo	/5			
		-6.43 a		-4.43 a					-0.410		-194.0 a		
		-6.37 a		-4.40 a					-0.210c		1-193.0 a		a -6.
		-3.55 c		-2.53 a					+0.205¢		-141.0 c	-33.1	c -2.
]			-2.67 c				 	+0.376¢				
				-1.14 c					+0.324 c				
				+0.533c					+0.163c				
				+0.533c					+1.43 a		-41.1 a		ļ
				+0.156c			\bot		+0.163c		+3.93c		
				+1.55 c						-1.22c	-7.30c		ļ
				-2.23 c						-3.23c	-47.0 c		
			<u> </u>	-10.7 c			-			-7.46c	+16.6 c		
	222 0 =			+1.56 b	1177 0	1/27 2			+1.61 a			1000	
	-737,0 c -161,0 a			-1.30 c +0.139c		+567.0 +66.5	C		+4.43 c +1.72 a	-5.35c		+37.3	

idence level is equal to or greater than 60% but less than 90° Standard error of estimate, N - Number of tests per regression analysis, I - Intercept value for regression equat

ration

┡											τ		, – –	
	77.00	СР	l Mo			Ce	HE	Ta	W	Re	R	SE	N	ı
1	Zr	СБ	MO	La]	ı a	-	Re	J	1 - E	لــــــــــــــــــــــــــــــــــــــ	
c	+61.6 c	-1.54c	-3.09		1+3	6.3 a	[-0.943c	-0.996c		0.560	26.5	111	32.5
c	+267.0 a	+2.93c	. +0.255			9.4 a	1	-0.002c	+0.600c		0.562		111	74.3
C	+127.0 €	+12.2 a	+3,53 a	ו	+5	3.6 c		+3.08 a	+4.42 a		0.752	23.5	108	75.0
		-320.0 c							+10.6c c		0.85	10.0	20	145.0
		-2352.0 c		:					-8.49 c		0.87	35.7	12	1742.0
		-2522.0 c		r					+9.20 c		0.89	15.7	20	1280.0
		-1511.0 c							-0.785c		0.85	26.3	16	1026.0
		-1925.0 c							-1.40 c		0.90	18.3	23	1012.0
		+305/1.0 c							-0.940c		0.92	17.3		-1422.0
	+705.0 c		-3.55				ļJ	+9.32 c	-9.91 c		0.97	0.98	31	-265.0
_	+2731.3 c		-1/1. 1	-51.7				+62.4 c	-174.7 c		1.00	0.94	13	
	+151.0 c							-2.97 c	-1.11 c	+0.200c		7.40	46	13.8
	<u>₩5,5</u> c							-0.831c				7.49	42	-63.4
b	+21.0 c	-1,32c						-0.374c	±1.61 c	+2.83 c		2.47	25	5.6
	-974.0 a	-13.7 c	1 44.06	<u> </u>		0.6560		-40.4 c	+2.45 c	L	0.91	29.9	42	-197.0
											!			
a	-56.C a			-29/10.0	a -2	6.6 a		-2.15 a	-2.00 a		0.89	13.4	48	127.0
a	-56.) a		+4.00 1		a -2	5.8 a		-2.05 a	-2.05 a		0.86	15.6	48	126.0
Ç*,	-33.1 c	-2.13c	+19.9 ;	-2243.0	c -1	j.1 c		-1.27 c			0.32	21.7	49	76.5
C								-2.75 c	-0.654c		0.94	3.67	11	24.6
cc a								-0.318c			0.57	5.95	15	-64.9
a								+1.91 b			0.99	1.29	11	-25.9
a								-2.06 b			0.85	3.31	28	-26.8
3 न								-0.647c	-0.762c		0.97	0.727	11	0.10
)c			I					+0.131c	+1.47 c		0.98	0.635	8	-132.0
C								-1.88 c	0.791c		0.96	3.04	16	+318.0
De e e								-5. 2 9 c	-14.8 c		0.998	1.72	8	+931.0
r,								+0.010c	+4.97 a		0.97	2.61	20	-98.5
C	###2.0 ··						-272.0c	+1.06 c	+1.46 c		0.95	3.66	55	+58.7
_	-37.3 .	J					-51.3a	-2.05 b	+1.87 c		0.82	3,16	42	-17.8

or regression equation

Table 8-A
Comparison of Signs of Regression Coefficients of RL
Nickel Alloys as a Function of Temperature

Time	Sea Salt								E1	eme	nts						
hour	ppm	Alloy Series	В	С	A1	Si	Ti	Cr	Рe	Со	Y	Zr	Cb	Mo	Ce	Ta	W
		Sur	fac	e L	088												
100	200	RL	+c	-c	+a	-c	-b	- a	-c	-c	-c	-c	-c	-c	-c	-c	-c
100	200	RL	+c	+c	+a	+c	40	- a	-c	- c	-c	-c	+c	+c	-c	+c	+c
100	200	RL	+c	+c	+c	-c	-c	- a	-c	-c	-c	-c	+a	+0	-c	+c	+a
		Maximu	n Pe	ene	tra	tion	2										
100	200	RL	-c	-c	+a	-c	- a	- a	-0	-c	+c	+c	 - c	-c	+a	-c	-c
100	200	RL	- a	+¢	+a	+0	+c	- a	-c	-c	+c	+a	+c	+0	+a	-c	+c
100	200	RL	-с	+c	-c	+ b	-b	- a	-0	-c	+c	+c	+a	+a	+c	+a	+a
	100 100 100 100	hour ppm 100 200 100 200 100 200 100 200	Nour ppm Alloy Series Sur	Nour ppm Alloy Series B	Nour ppm Alloy Series B C	Nour ppm Alloy Series B C Al	Nour ppm	Nour ppm Alloy Series B C Al Si Ti	Nour ppm	Nour ppm Alloy Series B C Al Si Ti Cr Fe	Nour ppm	Nour ppm Alloy Series B C Al Si Ti Cr Fe Co Y	Nour ppm	Nour ppm	Nour ppm Alloy Series B C Al Si Ti Cr Fe Co Y Zr Cb Mo	Nour ppm	Nour ppm Alloy Series B C Al Si Ti Cr Fe Co Y Zr Cb Mo Ce Ta

a ~ 95% confidence level

Table 9-A Comparison of Signs of Regression Coefficients of MELNI Alloys as a Function of Temperature

Temperature °F	Time hour		Alloy Series	Elements B C Al Ti Cr Fe Co Y Zr Cb Mo La Ta W Re
			Sur	face Loss
1600	1000	5	MELNI	+c +c -c +c +c +c -c -c +c +c +c +c +c +b +c
1800	1000	5	MELNI	+0-0+0+0-0+0+0+0+0+0+0+0+0+0
			Maximu	m Penetration
1600	1000	5	MELNI	-c +c -c -c +c +c +c +c +c -c -c +c +c
1800	1000	5	MELNI	+c -c -b -c +c +c +c -b +c -c +c -c +c +c

a - 95% confidence level

b - 90% confidence level

c - Confidence level is equal to or greater than 60% but less than 90%

b - 90% confidence level

c - Confidence level is equal to or greater than 60% but less than 90%

Table 10-A Comparison of Signs of Regression Coefficients in RL and MELNI Alloys at 1750° F for 100-Hour Test

Temperature °F	Time hour		Alloy Series	В	С	Al	Ti			ieni Co		Zr	Мо	Тa	W
			Surface 1	Loss	3										
1750	100	200	RL Nickel	+c	+c	+a	+c	- a	- c	-c	-c	-c	+c	+c	+c
1750	100	200	MELNI	ф ф	-c	+c	+c	+c	-c	+a	+b	+c	-b	+c	c
			Maximum Pen	etra	atio	<u>on</u>									
1750	100	200	RL Nickel	- a	+c	+a	+c	- a	-c	-c	+c	+a	+c	-c	+c -c
1750	100	200	MELNI	+ c	-c	+c	+ c	+c	-c	+a	+c	+c	-с	+c	-c

a - 95% confidence level

Table 11-A Comparison of Signs of Regression Coefficients of MELNI 1-13 at 1750° F as a Function of Exposure Time

Temperature	Time	Sea Salt							Ele	emei	nts					
°F	hour	ppm	Alloy Series	В	C	Al	Ti	Cr	Fe	Со	Y	Zr	Мо	La	Та	W
			Surfac	e L	oss						-					
1750	100	200	MELNI	+c	-c	+c	+c	+c	-c	+a	ď+	+c	-b	-c	+c	-c
1750	1000	5	MELNI	+c	-c	+c	+ c	+c	-c	+c	+c	+c	-c	-c	+c	-c
			Maximum P	ene	tra	ior	1									
1750	100	200	MELNI	+c	-c	+c	+c	+ c	- C	+a	+c	+c	-c	-c	+c	-c
1750	1000	5	MELNI	+c	-c	+c	+ c	+c	- c	+c	+c	¥	- C	ď	+c	-c

a - 95% confidence level

b - 90% confidence level

c - Confidence level is equal to or greater than 60% but less than 90%

b - 90% confidence level c - Confidence level is equal to or greater than 60% but less than 90%

Table 12-A Comparison of Signs of Regression Coefficients of Simple and Complex Nickel Alloys at 1750° F for 100 Hours

Temperature	Time	Sea Salt		·			Ele	emer	nts		
°F	hour	ppm	Alloy Ser	ies	C	Al	Тı	Cr	Co	Мо	W
		Sui	face Loss	i							
1750	100	800	RL		+c	+a	+c	- ā	- C	+0	+c
1750	100	200	TEL		- c	- C	- C	- a	+c	+c	-c
1750	100	200	MELNI		-c	÷c	÷c	+c	+ a	- b	-c
		Maxim	ım Penetra	itio	<u>n</u>						
1750	100	200	RL		+c	+a	+ c	- a	- C	+ĉ	+c
1750	100	200	TEL		- C	- C	-c	- 2	+c	+c	-c
1750	1 00	200	MELNI		- c	+c	+c	+c	+a	- C	-c

a - 95% confidence level b - 90% confidence level

Table 13-A Comparison of Signs of Regression Coefficients of TEL Alloys and Commercial Alloys at 1600° F for 500 Hours

Temperature °F	Tim∈ hour	Sea Salt ppm	Alloy	Series	С	Αl			nen Co		Мо	W
		St	irface	Loss								
1600	500	5 5	TEL	1-12	- a	+a	+c	- a	- C	- c	+c	-c
1600	500	5	сом		+c	+c	- C	+ c	- b	- C	+c	-c
		Maxir	num Per	netratio	on							
1600	500	5	TEL	1-12	-b +c	+a	+ b	- a	-c	- 0	+b	+c
1600	500	E _S	COM		+c	⊹ c	-0	¥	-b	-c	+c	+0

c - Confidence level is equal to or greater than 60% but less than 90%

a - 95% confidence level b - 90% confidence level

c - Confidence level is equal to or greater than $60\%\,\mathrm{but}$ less than 90%

Table 14-A Comparison of Signs of Regression Coefficients of RL Cobalt Alloys as a Function of Temperature

Temperature °F	Time hour	Sea Salt ppm	Alloy	Series	C	Ti	Сr	Ni			ment Cb		La	10	Тa	W
			St	urface I	los	5										
1675	100	500] 1	RL	+b	ı					- a	ı	i .	1		
1750	100	500	1	RL	+c		l .			•	- a		1		4	
1900	100	200]	RL	+c	- c	- a	+c	- C	-c	- C	+a	- C	- C	- C	+c
			Maxim	num Pene	etra	atio	on									
1675	100	200	1	RL	+a	- a	- a	-c	- a	- a	- a	- C	- a	- a	- a	- a
1750	100	200	1	ST	+b	- a	- a	- c	- a	- a	- a	+ b	~ (1	- a	- a	- a
1900	100	200	I	ST	+c	- C	- a	+c	-c	- C	- C	÷a	- C	- C	- C	+c

a - 95% confidence level b - 90% confidence level c - Confidence level is equal to or greater than 60% but less than 90%

Table 15-A

Comparison of Signs of Regression Coefficients in Simple and Complex Cobalt Alloys at Two Temperatures in 100-Hour Tests

Carbon

		face	Maxim		Concentra-
Alloys	1750° F	1900° F	Penetra 1750° F		tion Range w/o
Simple	+c	+c	d+	+c	0.01-0.45
Complex	+c	+c_	-c	+c	0.23-0.42

Chromium

		Suri	face	Maxim	num	Concentra-
į		Los	3 S	Penetra		tion Range
ì	Alloys	1750° F	1900° F	1750° F	1900° F	w/o
	Simple	-a	-a	-a	-a	15.0-35.0
	Complex	+c	+c	- c	-c	27.8-34.7

Nickel

	-			•	-//				
	Su	rí	ace		Maxi	mum	70	Concentra-	
	Loss			Penetration			tion Range	e	
Alloys	1750°	F	1900°	F	1750° F	1900° I	7_	w/o	
Simple	-c		+c	•	-c	+c	C	-10.2	
Complex	+c	ļ	-c		+c	+c	İc	5-10.7	ı

Yttrium

	TCCTION										
]	Sur	face	Maxim	num	Concentra-						
	Lo	SS	Penetr:	ation	tion Range						
Alloys	1750° F	1900° F	1750° F	1900° F	w/o						
Simple	a	-c	-a	-c	0-0.150						
Complex	-c	-a	- c	-c	0-0.190						

Tantalum

	Su	r f	ace		Ma	кir	num		Concentra-		
{	1	Loss			Penetration			tion Range			
Alloys	1750°	F	1900°	F	1750°	F	1900°	F	w/o		
Simple	-a		- c		- a		-c		0-15.0		
Complex	-0		+¢		-c		-c		0-4.0		

Tungsten

		1 411	93001		
	Sur	face	Maxim	num	Concentra-
	Lo		Penetra	ation	tion Range
Alloys	1750° F	1900° F	1750° F	1900° F	w/o
Simple	-a	+c	- a	+c	0-15.0
Complex	+c	+c	_c	+c	6.30-8.25

Note: Simple alloys are RL 10-15, 52-64 Complex alloys are MELCO 1-11

Table 16-A Comparison of Signs of Regression Coefficients of MELCO 3, 4, 5, 7-11 as a Function of Temperature

Temperature °F	Time hour	Sea Salt ppm	Alloy Series	В	С			nent	s	Та	W
Surface Loss											
1600	500	5	MELCO	+c	+c	+c	+c	+c	+c	+c	+c
2050	500	5	MELCO	+c	+C	- C	-c	+c	-c	+c	-c
2125	500	5	MELCO	+c	- C	- C	- C	- C	+c	- c	-c
		Maxim	mum Penetratio	on							
1600	500	5	MELCO	+c	-c	+c	+0	-c	-c	+c	+c
2050	500	5	MELCO	+c	-0	-c	-c	-c	-с	-c	-c
2125	500	5	MELCO	+c	+ C	- c	- C	-c	+ c	-c	- C

a - 95% confidence level

Table 17a-A
Comparison of the Regression Coefficients of Boron
in MELCO 1-11 at Two Temperatures and Two Test Conditions*

[Surfac	e Loss	[Maximum P	enceration
	1750° F	1900° F	Į	1750° F	1900° F
100 hours	+465.0	+556.0		+796.0	+1094.0
200 ppm	C	c		c	С
1000 hours	-163.0	+445.0		+76.0	+1320.0
5 ppm	c	С		С	a

Cencentration range of B is 0" to 0.015%.

b - 90% confidence level

c - Confidence level is equal to or greater than 60% but less than 90%

Table 17b-A

Comparison of the Regression Coefficients of Carbon in MELCO 1-11 at Two Temperatures and Two Test Conditions*

		Surface	e Loss	ł	Maximum P	enetration
		1 7 50° F	1900° F	1	1750° F	1900° F
:	100 hours	+1.30	+8.46		-10.4	+29.2
i	200 ppm	С	c		С	С
1	1000 hours	-8.01	-15.1	j	-7-79	+15.6
	5 ppm	c	c		С	c

^{*}Concentration range of C is 0.23% to 0.42 %.

Table 17c-A

Comparison of the Regression Coefficients of Chromium in MELCO l-ll at Two Temperatures and Two Test Conditions*

	Surface			enetration
	1 7 50° F	1900° F	1750° F	1900° F
100 hours	+0.213	+0.417	-2.67	-1.14
200 ppm	c	С	С	С
1000 hours	+0.021	-0.942	+0.533	+0.583
	c	ъ	c	c

^{*}Concentration range of Cr is 27.3 to 34.7%.

Table 17d-A

Comparison of the Regression Coefficients of Nickel in MELCO 1-11 at Two Temperatures and Two Test Conditions*

[Surfac	.		enetration
	1 7 50° F	1900° F	1 7 50° F	1900° F
100 hours	+0.544	-0.207	+0.376	+0.324
200 ppm	C	C	(*	C
1000 hours	+0.030	+1.30	+0.163	+1.43
5 ppm	c	a	e	a

Table 17e-A

Comparison of the Regression Coefficients of Copper in MELCO 1-11 at Two Temperatures and Two Test Conditions*

	Surface		1	Maximum Pe	enetration
	1750° F	1900° F		1 7 50° F	1900° F
100 hours	-2.12	+0.058		-2.87	-1.35
200 ppm	С	c		С	С
1000 hours	-0.177	-3. 55		+1.43	-2.87
5 ppm	С	a		С	ď

^{*}Concentration range of Cu is 0% to 2.95%.

Table 17f-A

Comparison of the Regression Coefficients of Yttrium in MELCO 1-11 at Two Temperatures and Two Test Conditions*

[Surface	e Loss]	Maximum P	enetration
	1750° F	1900° F] .	1750° F	1900° F
100 hours	-0.323	-33.0		-10.4	-27.0
200 ppm	c	a		C	c
1000 hours	+5.01	+25.6		-43.6	-41.1
5 ppm	С	b		a	a

^{*}Concentration range of Y is 0% to 0.190%.

Table 17g-A

Comparison of the Regression Coefficients of Tantalum in MELCO 1-11 at Two Temperatures and Two Test Conditions*

į.	Surfac		[Maximum P	enetration
	1750° F	1900° F		1750° F	1900° F
l 100 hours	-1.66	+0.437		-2.75	-0.313
200 ppm	c	С		c	i c
1000 hours	+0.077	-2.31		+1.91	-2.06
5 ppm	c	a		b	: b

^{*}Concentration range of Ta is 0% to 4.00%.

Table 17h-A
Comparison of the Regression Coefficients of Tungsten
in MELCO 1-11 at Two Temperatures and Two Test Conditions*

1	Surface Loss			Maximum Penetration				
	1750° F	1900° F]	1750° F	1900° F			
100 hours	+1.41	+0.965		-0.654	+2.77			
200 ppm	c	С		С	С			
1000 hours	+0.001	-2.83		+2.79	+0.460			
5 ppm	c	a		С	С			

^{*}Concentration range of W is 6.30% to 8.25%.

Table 18-A
Hot-Corrosion Results for Factorial Alloys¹
(Nickel-Base, 100 Hours of Test Operation)

	,	Chem			(2) Maximum Penetration, mils									
RL)	Allo		_		1675° F			1750° F			1900°	F	
Alloy		9	6						Test ³					
No.	N1	Cr	Al	Ti	2	3	4	7	8	9	12	13	14	
44	Bal	15.0	2.5	4.5	54.6	97.9	10.7	36.6	3 9.9	50.1	34.9	22.2	58.6	
45	Bal	15.0	2.5	9.0	ვ.6	3.9	7.5	15.3	51.1	11.4	11.5	12.2	21.6	
46	Bal	15.0	5.0	4.5	15.8	>130.0	5.9	10.3	>130.0	43.7	46.7	21.5	63.3	
47	Bal	15.0	5.0	9.0	7.7	9.9	6.5	9.4	>130.0	ક.9	44.5	15.5	56.4	
48	Bal	20.0	2.5	4.5	12.5	13.5	9.0	11.5	17.9	12.6	17.0	12.4	33.4	
49	Bal	20.0	2.5	9.0	7.1	7.0	6.6	9.6	3.0	3.7	13.6	16.3	14.1	
50	Bal	20.0	5.0	4.5	8.6	7.7	7.9	10.0	10.4	23.7	27.1	23.3	45.5	
51	Bal	20.0	5.0	9.0	6.6	5.4	6.1	3.2	>130.0	8.9	31.1	11.3	14.4	

Temperature - as given; Time - 100 hours; Fuel - diesel (1% sulfur); Air/Fuel - 30/1; Sea Salt - 200 ppm of air; Specimen Size - approximately 0.130 inch in diameter by 1.25 inches in length.

A maximum penetration >130 mils indicates that the specimen has been corroded all the way through.

³Each test number represents a separate test run in which one specimen of each alloy is placed in a hot corrosion test chamber, which is then operated under specified conditions. Bal - Balance.

Table 19a-A
Analysis of Variance for
Hot-Corrosion Results for Factorial Alloys
(Nickel-Base, 100 Hours of Test Operation, All Data Included)

Source of Variance	Sums of Squares	Degrees of Freedom	Mean Squares	'F' Ratio
Temperature (T)	2378.39	2	1189.28	1.45
Chromium (Cr)	6367.56	1	6367.56	7.78*
Aluminum (A1)	1833.15	1	1833.15	2.24
Titanium (Ti)	2704.80	1	2704.80	2.31
Residual	<u>53998.18</u>	<u>66</u>	818.15	
Total	67282.08	71		

^{*}Significant at the 1% level.

Table 19b-A
Analysis of Variance for
Hot-Corrosion Results for Factorial Alloys
(Nickel-Base, 100 Hours of Test Operation
Test Runs 3, 8, and 13 Eliminated)

Source of Variance	Sums of Squares	Degrees of Freedom	Mean Squares	'F' Ratio
Temperature (T)	2033.32	2	1016.66	12.31*
Chromium (Cr)	826.75	1	823.75	10.00*
Aluminum (Al)	19.93	1	19.93	<1.00
Titanium (Ti)	1007.06	1	1007.06	12.24*
Residual	1438.99	<u>18</u>	82.72	
Total	5376.05	23		

^{*}Significant at the 1% level.

Note: The following test results have been averaged in Table 19b for each alloy: Tests 2 and 4, 7 and 9, and 12 and 14.

Table 20-A

Hot-Corrosion Results for Factorial Alloys¹
(Cobalt-Base, 100 Hours of Test Operations)

	1	Chem			Maximum Penetration, mils							
DISCO	by 4			1675° F		O° F	1900° F					
Alloy			Tes	Test 1 ⁽²⁾		Test 2 ⁽²⁾		t 3 ⁽²⁾				
No.(2)	Со	Cr	Ni	Ta	A	В	A	В	A	В		
1	Bal	25	5	6	3.9	2.8	6.5	8.6	14.7	16.6 ⁽³⁾		
2	Bal	30	5	6	5.3	3.8	6.8	9.2	14.9	18.0		
3	Bal	35	5	6	5.4	8.3	8.2	9.9	13.9	16.1		
4	Bal	25	15	6	4.8	5.0	7.2	7.6	14.3	17.2		
5	Bal	30	15	6	6,8	5.8	11.3	8.6	15.6	17.1		
6	Bal	3 5	15	6	6.6	8.9	8.9	12.4	14.8	16.2		
7	Bal	25	25	6	5.9	5.0	8.4	9.6	16.5	18.7		
8	Bal	30	25	6	6.6	5.1	10.7	11.8	18.6	19.7		
9	Bal	3 5	25	6	6.8	10.3	12.1	13.6	16.1	20.7		

¹Temperature - as given; Time - 100 hours; Fuel - diesel (1% sulfur); Air/Fuel - 30/1; Sea Salt - 200 ppm of air; Specimen Size - approximately 0.130 inch in diameter by 1.25 inches in length.

²DISCO 1A to 9A - Specimens are used, as cast. DISCO 1B to 9B - Specimens are heat treated, 1200° C (2192° F), 14 hours, WC. Heat-treated (B) and nonheat-treated (A) specimens of each allow were tested at one time, under a specified temperature condition.

³Estimated value.

Table 21a-A Analysis of Variance for Hot-Corrosion Results for Factorial Alloys (Cobalt-Base, Overall Factors)

Source of Variance	Sums of Squares	Degrees of Freedom	Mean Squares	'F' Ratio Line	Line
DISCO, A vs B(H)	18.96	1	18.96	1/5 = 3.09	1
Temperature (T)	1097.21	2	548.60	2/5 = 89.35*	5
Chromium (Cr)	43.34	, 2	21.67	3/6 = 3.74	3
Nickel (Ni)	54.72	2	27.36	4/8 = 28.21*	Į‡
HxT	12.23	2	6.14	5/8 = 6.33*	5
H x Cr	14.26	2	7.13	6/8 = 7.35*	6
TxCr	26.11	4	6.53	7/8 = 6.73*	7
Residual	<u> 36.93</u>	<u>38</u>	0.97		8
Total	1303.81	53			

^{*}Significant at the 1% level (very significant).

'F' - Fisher's ratio.

Table 21b-A Analysis of Variance for Corrosion Results for Factorial Alloys (Cobalt-Base, Over 3-Temperature Levels)

Source of Variance	Sums of Squares	Degrees of Freedom	Mean Squares	'F' Ratio Line	Line
	Te	emperature -	1675° F		
DISCO, A vs B(H) Chromium (Cr) Nickel (Ni) H x Cr Residual Total	0 40.30 3.63 19.12 4.79 72.43	1 2 2 2 10	0 20.20 1.5 ^h - 9.50 0.43	1/4 = 0 2/4 = 2.11 5/5 = 9.04* 4/5 = 19.91*	1 2 3 4 5
	Te	emperature -	1750° F		
District And B(H) Clinous up to ri Niclinous And Residual Total	7. 35 24.41 15.35 77.27	1 () 1() 17	6,46 12.55 12.20 1.32	1/4 = 4.27** 2/4 = 9.49* 3/4 = 9.24*	1 2 3 4
	Te	emperature -	1903° F		
Disc, A vs B(H) Chromium (Cr) Nickel (Ni) Fesi bal Total	04.02 4.01 07.15 (4.7) (1.35)	1 2 7 10 17	24.22 2.00 15.56 0.50	1/4 = 43.44* 2/4 - 4.00** 3/4 = 27.12*	1 2 3 4

^{*} institute the l'level (very significant).

Appendix B

Technical References

- 1 Shepard, M. L., N. J. Olson, and M. J. Donachie, "Hot Corrosion Studies of Uncoated Experimental Nickel-Base Superalloys," Proceedings of the Air Force Materials Laboratory Fiftieth Anniversay Technical Conference on Corrosion of Military and Aerospace Equipment, Denver, Colo, 23-25 May 1967, Air Force Rept AFML-TR-67-329, Nov 1967, p. 1455
- 2 Ryan, K. H., J. R. Kildsig, and P. E. Hamilton, "Investigation of Hot Corrosion of NIckel-Base Superalloys Used in Gas Turbine Engines," Air Force Rept AFML-TR-67-306, Air Force Contract AF33(615)-5211 with Allison Div, General Motors, AD 825-297, Aug 1967
- 3 Bergman, P. A., C. T. Sims, and A. M. Beltran, "Development of Hot Corrosion Resistant Alloys for Marine Gas Turbine Service," MEL Sponsored Rept 131/66, Final Report under Contract N600(61533)63218 with the General Electric Co, AD 629-786, 21 Jan 1966
- 4 Gambino, J. R., et al, "Hot Corrosion Mechanism Studies," MEL Sponsored Rept 132/66, Final Report under Contract N600(61533)63219 by the General Electric Co, AD 629-598, 15 Feb 1966
- 5 Seybolt, A. U., P. A. Bergman, and M. Kaufman, "Hot Corrosion Mechanism Study, Phase II," MEL Sponsored Rept 2591, Final Report under Contract N600(61533)65595 with the General Electric Co, AD 825-791, 31 Oct 1967
- 6 Bergman, P. A., A. M. Beltran, and C. T. Simms, "Development of Hot-Corrosion-Resistant Alloys for Marine Gas Turbine Service," MEL Sponsored Rept 2437, Final Report under Contract N600(61533)65661 with the General Electric Co, AD 825-801, 1 Oct 1967
- 7 Bergman, P. A., C. T. Sims, and A. M. Beltran, "Development of Hot-Corrosion-Resistant Alloys for Marine Turbine Service,"

 Hot Corrosion Associated with Gas Turbines, ASTM STP 421,

 1967, pp 38-63
- 8 Hardt, R. W., J. R. Gambino, and P. A. Bergman, "Marine Hot Corrosion Mechanism Studies," Hot Corrosion Problems
 Associated with Gas Turbines, ASTM STP 421, 1967, pp 64-83
- 9 Hoel, P. G., Introduction to Mathematical Statistics, New York, N.Y., John Wiley and Sons, Inc., 1947
- 10 Brownlee, K. A., <u>Industrial Experimentation</u>, Brooklyn, N.Y., Chemical Publishing Co, Inc., 1949

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The use of gas turbines in marine	power plan	nts depen	ds in part on the
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ranged from 1600° to 2125° F. Time:			
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million calt Corrosion was measure	ed by reco	rding bot	h surface loss

and maximum penetration. This experimental work was performed by the General Electric Company under contract to the Naval Ship Research and Development Center. For each group of alloys tested under similar conditions, a linear regression equation was found that shows the average contribution of each alloying element to the amount of corrosion. The effects of the alloying elements were found to vary with changes in temperature, salt concentration, and whether or not the particular element was part of a simple binary or tertiary alloy, or a complex

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alloy. Analyses of variance methods were applied to two sets of factorially designed compositions, one of nickel-base alloys and one of cobalt-base alloys, to determine the possible significance on corrosion of various proportions of single elements and interactions among elements. It was found that in the cobalt alloys significant interactions existed between heat treatment and temperature as well as between heat treatment and chromium content.

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